

The nature of the CO₂-concentrating mechanisms in a marine diatom, *Thalassiosira pseudonana*

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Summary

- Diatoms are widespread in aquatic ecosystems where they may be limited by the supply of inorganic carbon. Their carbon dioxide-concentrating mechanisms (CCMs) involving transporters and carbonic anhydrases (CAs) are well known, but the contribution of a biochemical CCM involving C₄ metabolism is contentious.
- The CCM(s) present in the marine-centric diatom, *Thalassiosira pseudonana*, were studied in cells exposed to high or low concentrations of CO₂, using a range of approaches.
- At low CO₂, cells possessed a CCM based on active uptake of CO₂ (70% contribution) and bicarbonate, while at high CO₂, cells were restricted to CO₂. CA was highly and rapidly activated on transfer to low CO₂ and played a key role because inhibition of external CA produced uptake kinetics similar to cells grown at high CO₂. The activities of phosphoenolpyruvate (PEP) carboxylase (PEPC) and the PEP-regenerating enzyme, pyruvate phosphate dikinase (PPDK), were lower in cells grown at low than at high CO₂. The ratios of PEPC and PPDK to ribulose biphosphate carboxylase were substantially lower than 1, even at low CO₂.
- Our data suggest that the kinetic properties of this species results from a biophysical CCM and not from C₄ type metabolism.

Introduction

Diatoms are unicellular microalgae that appeared *c.* 120–250 million yr ago (Sims *et al.*, 2006; Sorhannus, 2007) and have since evolved to form a group of 30 000–100 000 species (Mann & Vanormelingen, 2013) that are ubiquitous in aquatic and moist habitats. Like other Chromista, diatoms are thought to be derived from endosymbioses between a heterotrophic cell, a red alga, and possibly a genetic contribution from a green alga (Armbrust, 2009; Moustafa *et al.*, 2009; Deschamps & Moreira, 2012). Because of their complex evolutionary history, the diatom genome comprises genes from algae, as well as animals and bacteria. This confers diatoms with features, such as the presence of the urea cycle, that differentiate them from other photoautotrophs (Allen *et al.*, 2011). This biochemical diversity could be linked to their ecological success, as the dominant oceanic phytoplankton switched from cyanobacteria and green algae to Chromista, such as diatoms and haptophytes, (Falkowski *et al.*, 2004) at a time when atmospheric CO₂ concentration declined and O₂ concentration increased (Katz *et al.*, 2005; Armbrust, 2009; Raven *et al.*, 2012). Today, diatoms are responsible for up to 40% of primary production in the Earth's largest ecosystem, the ocean (Roberts *et al.*, 2007a), and a large proportion of the export of organic carbon to the ocean floor (Sarhou *et al.*, 2005).

Ribulose-1,5-biphosphate carboxylase/oxygenase (Rubisco) is universally present in photosynthetic organisms and catalyses two

reactions, a carboxylation of ribulose-1,5-bisphosphate (RuBP) with CO₂, and an oxygenation of RuBP with O₂ (Bowes *et al.*, 1971; Gontero & Salvucci, 2014). These two reactions compete and thus the oxygenase reaction is favoured at low CO₂ concentrations, reducing photosynthesis (Badger *et al.*, 1998). The Michaelis–Menten constant (K_m) for CO₂ of Rubisco of diatoms varies from 20 to 60 μM, which is higher than the CO₂ concentration in marine ecosystems at equilibrium with the current atmosphere of 400 ppm (*c.* 16 μM depending on temperature; Badger *et al.*, 1998; Whitney *et al.*, 2011). To circumvent or reduce carbon limitation of photosynthesis, some aquatic photosynthetic organisms, including diatoms, possess carbon dioxide-concentrating mechanisms (CCMs) that elevate the CO₂ concentration around Rubisco, thus decreasing the oxygenase reaction and thereby increasing the rate of photosynthesis (Roberts *et al.*, 2007a).

Several types of CCM are known, based on biophysical or biochemical processes. Biophysical CCMs involve active transport of CO₂ or bicarbonate (HCO₃[−]), and are present in many diatoms (Matsuda *et al.*, 2011). For instance, in marine diatoms, the SLC4 HCO₃[−] transporter is present in *Phaeodactylum tricoratum* (Nakajima *et al.*, 2013), and homologous encoding genes are also found in *Thalassiosira pseudonana* (Armbrust *et al.*, 2004). Carbonic anhydrase (CA) maintains equilibrium between CO₂ and HCO₃[−] by catalysing the reversible interconversion of CO₂ and water into HCO₃[−] and protons. It plays a role in

diatom CCMs (Hopkinson *et al.*, 2011; Matsuda *et al.*, 2011) and its expression is increased under a low CO₂ concentration in *P. tricornutum* and *T. pseudonana* (Harada *et al.*, 2005; Crawford *et al.*, 2011; Hopkinson *et al.*, 2013). In *P. tricornutum*, some CAs are redox-regulated and activated by reduced thioredoxins, suggesting that they are active during the day and inactive at night, which is consistent with their participation in a CCM (Kikutani *et al.*, 2012).

Biochemical CCMs involving C₄-type photosynthesis have been suggested to be involved in CO₂ assimilation in some diatoms (Reinfelder *et al.*, 2000). A functional C₄ CCM requires an additional carboxylation enzyme, typically phosphoenolpyruvate carboxylase (PEPC), which catalyses the carboxylation of phosphoenolpyruvate (PEP) with HCO₃⁻, forming a C₄ carbon compound. This compound is then cleaved by one of three decarboxylating enzymes to produce CO₂ in the vicinity of Rubisco (Sage, 2004). Although C₄ metabolism in terrestrial plants is usually associated with Kranz-type anatomy (Sage, 2004), some terrestrial plants, such as *Borszczowia aralocaspica*, perform C₄-type photosynthesis within one cell (Voznesenskaya *et al.*, 2001). Similarly, in aquatic environments, *Hydrilla verticillata*, *Ottelia alismoides*, *Egeria densa*, *Udotea flabellum* and *Ulva lynza* are believed to perform this type of photosynthesis without Kranz anatomy (Reiskind & Bowes, 1991; Magnin *et al.*, 1997; Lara *et al.*, 2002; Xu *et al.*, 2013; Zhang *et al.*, 2014) and so it is feasible that this pathway may be present in diatoms (Kroth, 2015).

In two diatoms whose genomes are fully sequenced and annotated, *T. pseudonana* (Armbrust *et al.*, 2004) and *P. tricornutum* (Bowler *et al.*, 2008), all the genes required for C₄-type photosynthesis are present. Thus, diatoms have the genetic potential to operate a C₄ pathway. However, this possibility remains controversial (Raven, 2010) as there are a range of contradictory results for the possession of C₄ metabolism in diatoms based on different approaches such as ¹⁴C labelling, use of specific C₄ enzyme inhibitors (Reinfelder *et al.*, 2004), proteomic analysis, transcriptomic analysis, enzyme localization and RNA silencing (McGinn & Morel, 2008; Kustka *et al.*, 2014; Tanaka *et al.*, 2014). A recent study (Kustka *et al.*, 2014), however, reaffirmed the operation of C₄ photosynthesis in *T. pseudonana* grown at low CO₂ and concluded that the nature of the CO₂ delivery system to the chloroplast needs to be investigated further (Samukawa *et al.*, 2014).

The aim of this study was therefore to decipher the roles of biophysical and biochemical CCMs in a model diatom, *T. pseudonana*, using a range of techniques. We studied the effect of growth in air (400 ppm CO₂) and extremely high (20 000 ppm) and low (50 ppm CO₂) on growth rate, photosynthetic kinetics, the activity of CA and the enzymes involved in C₄-type metabolism.

Materials and Methods

Strain, media and culture condition

Thalassiosira pseudonana Hasle & Heim., strain CCAP 1085/12 (equivalent to CCMP1335, the strain whose genome has been sequenced), was grown in F/2 + Si medium, pH 8, in artificial

sea water (mM: 380 NaCl, 3 KCl, 4.39 CaCl₂, 1.71 NaHCO₃, 20.8 MgSO₄, 0.88 NaNO₃, 0.036 NaH₂PO₄, 0.11 Na₂SiO₃), trace elements (µM: 12.3 Na₂EDTA, 11.7 FeCl₃, 40.1 CuSO₄, 0.077 ZnSO₄, 0.042 CoCl₂, 0.91 MnCl₂, 0.013 Na₂Mo₄) and vitamins (nM: 0.37 B12 (cyanocobalamin), 300 B1 (thiamine-HCl) and 2.05 B8 (biotin)).

Cultures were maintained in a growth cabinet (Innova 4230; New Brunswick Scientific, Edison, NJ, USA) at 16°C with continuous illumination at 50 µmol photon m⁻² s⁻¹ photosynthetically active radiation (PAR, spectral band 400–700 nm) measured with a 2π sensor (Q201; Macam Photometric, Livingstone, UK) and were constantly shaken at 90 rpm. The cultures were aerated with one of three gas mixtures (50, 400 or 20 000 ppm) at a gas flow rate of 130 ml min⁻¹ using mass-flow regulators (El-Flow; Bronkhorst High-Tech BT, Nijverheidsstraat, the Netherlands) that mixed air, 2% CO₂ in air, and air that had been passed through soda lime, to remove CO₂ (Intersurgical, Wokingham, UK). Dissolved CO₂ concentrations calculated using equations in Weiss (1974) were 2, 16 and 800 µM. Concentrations of CO₂ and other components of the carbonate system were calculated from pH, alkalinity, temperature and salinity using the dissociation constants in Goyet & Poisson (1989). During growth experiments, pH was checked daily using a combination pH electrode and meter (Eutech pH 2700; Eutech Instruments, Landsmeer, the Netherlands), optical density (OD) was measured at 600 nm using a Perkin Elmer Lambda 25 UV/VIS spectrophotometer (Waltham, MA, USA) and number of cells was counted by microscopy using a Neubauer chamber (Brand GmH, Wertheim, Germany). Growth rates were calculated as:

$$\text{Growth rate} = \left(\frac{\log_e(y_B) - \log_e(y_A)}{x_B - x_A} \right) \quad \text{Eqn 1}$$

where log_e(y_B) and log_e(y_A) correspond to the natural logarithm of OD at 600 nm or cell density (cell ml⁻¹) measured at the start and end of the exponential phase and x_B and x_A correspond to the time (day) of these two points.

Kinetics of O₂ evolution

Rates of net photosynthesis were measured as oxygen evolution in an electrode chamber thermostatted at 16°C (Oxygraph; Hansatech Instruments, Norfolk, UK) using O₂ View software (Hansatech, Norfolk, UK). The chamber was illuminated with a tungsten lamp with a hot-mirror cutoff filter at 750–1100 nm (HMC-1033; UQG (Optics), Cambridge, UK) to minimize heat input to the chamber. The cells received 200 µmol m⁻² s⁻¹ PAR, which preliminary experiments had shown to be saturating but not photoinhibiting. Cultures from the exponential phase were centrifuged at 3720 g for 10 min at 16°C (Beckman Coulter Allegra[®] X-15R Centrifuge (Pasadena, CA, USA); rotor: 4750A) and the pellet was rinsed twice, and resuspended in artificial sea water containing 10 mM HEPES at either pH 7 or pH 8. A suspension (1 ml) containing ± 20 million cells was added to the oxygen electrode chamber. Respiration was measured after 10 min in the dark to allow steady-state rates to be produced. The cells were

then illuminated and when net oxygen evolution had ceased, small volumes of stock (1, 10 and 100 mM) NaHCO₃ were added to produce a range of concentrations of dissolved inorganic carbon (DIC; 10, 20, 50, 100, 150, 200, 500, 1000 and 2000 μM) and the rate of change of oxygen concentration was recorded. To study the effect of CA on the rate of photosynthesis, 0.4 mM (final concentration) of acetazolamide (AZA; Sigma-Aldrich), an inhibitor of external CA, was added directly to the oxygen electrode in the light once oxygen evolution had ceased and immediately before the first addition of DIC. Biological duplicates and experimental triplicates were analysed, giving six replicates in total. The response of rate of net photosynthesis to the concentration of DIC was fitted to a slightly modified Michaelis–Menten equation that took into account the compensation point for DIC.

At pH 7, CO₂ represents 8% of DIC, while at pH 8 it only represents 0.8%. This difference was used to discriminate between the effects of CO₂ and HCO₃⁻ on net oxygen evolution using a model that assumes separate uptake of these two forms of inorganic carbon with different *K*_{1/2} and compensation concentrations but a common total maximum uptake rate:

$$\text{Net rate of photosynthesis} = \left(\frac{\alpha \times V_{\text{net}}^{\text{max}} \times (\text{CO}_2 - \text{CP}^{\text{C}})}{K_{1/2}^{\text{C}} + (\text{CO}_2 - \text{CP}^{\text{C}})} \right) + \left(\frac{(1 - \alpha) \times V_{\text{net}}^{\text{max}} \times (\text{HCO}_3^- - \text{CP}^{\text{B}})}{K_{1/2}^{\text{B}} + (\text{HCO}_3^- - \text{CP}^{\text{B}})} \right)$$

Eqn 2

where (rate as μmol O₂ mg⁻¹ Chl *a* h⁻¹ and concentration as μmol l⁻¹) *V*_{net}^{max} is the maximum rate of net photosynthesis; α is the proportion of *V*_{net}^{max} resulting from CO₂ uptake; CO₂ is the concentration of CO₂; CP^C is the CO₂ compensation concentration; *K*_{1/2}^C is the concentration of CO₂ yielding half-maximal rates of net photosynthesis; HCO₃⁻ is the concentration of HCO₃⁻; CP^B is the HCO₃⁻ compensation concentration; and *K*_{1/2}^B is the concentration of HCO₃⁻ yielding half-maximal rates of net photosynthesis.

The best fit of the model parameters to the data was obtained by minimizing the residual sum of squares of the difference between the measured and modelled rates of net photosynthesis.

Chlorophyll extraction and measurement

The culture was centrifuged at 3720 *g* for 10 min at 4°C. The pellet was rinsed in distilled water and re-centrifuged, and 2 ml of 96% ethanol was added. After incubation for 15 min at 4°C in the dark, the supernatant was removed and a second extraction was performed. The optical density of the bulked supernatant was measured with the spectrophotometer at 629 and 665 nm. Optical density at 750 nm was negligible and so uncorrected values were used to calculate concentrations of Chl *a* using the equation in Ritchie (2006):

$$\text{Chl } a \text{ (}\mu\text{g ml}^{-1}\text{)} = -1.4014 \times A_{629} + 12.1551 \times A_{665}. \quad \text{Eqn 3}$$

Protein extraction and content

The soluble protein extracts were prepared as described previously (Erales *et al.*, 2008; Mekhalfi *et al.*, 2014) in a buffer containing 1 mM NAD. The soluble protein concentration of crude extracts was assayed using the Bio-Rad (Hercules, CA, USA) reagent using BSA as a standard (Bradford, 1976).

Enzyme activity measurement

All enzyme activities were measured on cells from the exponential phase of growth. CA activity was measured spectrophotometrically using bromothymol blue as a pH indicator. Crude extracts of cells were incubated in 1.6 ml of buffer (25 mM Tris, 6.4 μM bromothymol blue at pH 9.1) in a cuvette at 3°C. The reaction was started by adding 0.4 ml of CO₂-saturated milliQ water that had been kept on ice. Blanks were performed for each assay by omitting the sample. CA activity was estimated from the time required for the disappearance of the bromothymol blue absorbance at 620 nm which corresponds to a pH decrease from 9.1 to 6.2. Enzyme activity was calculated in Wilbur-Anderson Units (WAU) using the equation (Wilbur & Anderson, 1948):

$$\text{WAU} = T_0 / T_1 - 1 \quad \text{Eqn 4}$$

where *T*₀ and *T*₁ correspond to the acidification time without (blank) and with the sample in the reaction mixture, respectively. External CA (eCA) was determined on intact cells; total CA was determined on cells that had been broken by sonicating (Erales *et al.*, 2008); internal CA (iCA) was calculated from the difference between total CA and eCA.

Other enzyme activities were measured from the rate of appearance or disappearance of NADH or NADPH at 340 nm at room temperature (20–25°C). All biochemicals were obtained from Sigma. PEPC, NAD-dependent malic enzyme (NAD-ME) and pyruvate phosphate dikinase (PPDK) activities were measured as described previously (Zhang *et al.*, 2014). The activities of fully CO₂-activated and nonactivated Rubisco were measured. To activate Rubisco, the extract was preincubated in 50 mM Tris, 0.1 mM EDTA, 15 mM MgCl₂, 40 mM bicarbonate and 5 mM dithiothreitol, pH 8.0, for 10 min before assay in a 1 ml cuvette. To measure activity, five units of phosphoglycerate kinase and glyceraldehyde-3-phosphate dehydrogenase, 1 mM ATP and 0.2 mM NADH were added and the reaction was started by adding 1 mM ribulose 1,5-bisphosphate. The activity of nonactivated Rubisco was measured as described earlier, but the reagents were all added at the same time without preincubation. The activation procedure is equivalent to that used to carbamylate Rubisco in higher plants, cyanobacteria and a range of algae, but whether or not this is the mechanism involved in activation has not, to our knowledge, been studied in diatoms.

Rubisco, PEPC and CA activities were measured as a function of time after the switch to low CO₂, and the curves were fitted with SIGMA PLOT software (Systat Software GmbH, Erkrath, Germany) to experimental data using Eqn 5 for the carboxylases, and their activity ratio and Eqn 6 for CA:

$$A = A_0 + p(1) \times (e^{-p(2) \times t}) \quad \text{Eqn 5}$$

and

$$A = A_0 + p(1) \times (1 - e^{-p(2) \times t}) \quad \text{Eqn 6}$$

where A is the rate of reaction mg^{-1} of Chl a , A_0 is the activity at the beginning of the experiment before the switch to low CO_2 , $p(1)$ is the amplitude and $p(2)$ is the time constant.

Statistical analysis

Results were analysed using SIGMA PLOT (v.11.0).

Results

Effect of CO_2 on growth rate

The growth rate of *T. pseudonana*, was determined at three concentrations of CO_2 . In the absence of algae, the pH values at equilibrium with 50, 400 and 20 000 ppm CO_2 were 8.8, 7.9 and 6.7, respectively. The corresponding calculated CO_2 concentrations were 1, 19 and 320 μM for 50, 400 and 20 000 ppm, respectively, which were similar to the theoretical concentrations apart from at the highest CO_2 concentration. In the cultures with algae at 20 000 ppm, the pH dropped quickly to 6.9 and then remained constant for several days (Fig. 1b). During exponential growth, the geometric mean pH was 6.95, equivalent to a CO_2 concentration of $c. 180 \mu\text{M}$. At 400 ppm, pH increased progressively during the exponential phase and reached up to pH 9–9.2 at the beginning of the stationary phase. The geometric average pH during the exponential phase was 8.55, which is equivalent to a CO_2 concentration of $c. 3 \mu\text{M}$. Similarly, when the cells were shifted from 20 000 to 50 ppm, the pH also increased and reached over 9.5, corresponding to $<0.1 \mu\text{M}$ CO_2 (Fig. 1b). These elevated pH values were caused by the rate of CO_2 consumption at high cell density exceeding the rate of CO_2 supply. The maximum specific growth rate ($0.70 \pm 0.01 \text{ d}^{-1}$) at 20 000 ppm CO_2 was $c. 1.3$ -fold higher than at 400 ppm CO_2 ($0.54 \pm 0.02 \text{ d}^{-1}$, Fig. 1a). A similar ratio of growth rate at high and ambient CO_2 was found, based on cell counts (data not shown). *T. pseudonana* was unable to grow when transferred from 20 000 to 50 ppm (Fig. 1a). As the pH in the 50 ppm treatment was stable for several days, it indicates that the treatment was not so severe as to cause cell death and this is consistent with the optical density data (Fig. 1a).

Photosynthetic activity

The maximal rate of net photosynthesis ($V_{\text{net}}^{\text{max}}$) of *T. pseudonana* grown under 400 ppm CO_2 and measured at pH 7 ($c. 110 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg}^{-1}$ Chl a) was similar to that measured at pH 8 (Fig. 2a; Table 1). By contrast, the half-saturation concentration for DIC at pH 7 was $c. \text{threefold}$ lower than that at pH 8 (Table 1). The DIC compensation point was also threefold lower at pH 7 than at pH 8. For cells grown at 20 000 ppm CO_2 , the

maximal photosynthetic activity ($V_{\text{net}}^{\text{max}}$) measured at pH 7 was $c. 205 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg}^{-1}$ Chl a , which was twice that measured at pH 8 (Fig. 2d, Table 1). The half-saturation concentrations for DIC at pH 7 and pH 8, by contrast, were rather similar and $c. 50 \mu\text{M}$. The DIC compensation points were 3 and 5 μM at the two pH values. When *T. pseudonana* was grown at 20 000 ppm CO_2 , the slope of rate of photosynthesis against DIC was between 3.5 and 6.5-fold lower than that found when *T. pseudonana* was grown at 400 ppm CO_2 (Table 1).

The different kinetic parameters at pH 7 and 8 are consistent with different proportions of CO_2 and HCO_3^- being present at these two pH values and we used this to develop a model that distinguished between CO_2 and HCO_3^- uptake (Eqn 2). For cells grown at 400 ppm, this model gave a good fit to the data (R^2 of 0.92; Table 2) and a $V_{\text{net}}^{\text{max}}$ of $112 \mu\text{mol O}_2 \text{ h}^{-1} \text{ mg}^{-1}$ Chl a , corresponding to the sum of CO_2 - and HCO_3^- -dependent uptake, which was similar to that found when modelling kinetics against DIC. The model predicted that at saturation, CO_2 contributed 70% and HCO_3^- contributed 30% to the maximal rate (Fig. 2b, c). The half-saturation concentration for CO_2 was 0.4 μM which was 7.5-fold lower than that for HCO_3^- at 3 μM . The compensation points were close to 0 for CO_2 and 7 μM for HCO_3^- . The slope of uptake was seven times higher for CO_2 than for HCO_3^- .

For cells grown at 20 000 ppm, the model fitted the data less well than at 400 ppm (R^2 of 0.60; Table 2). The $V_{\text{net}}^{\text{max}}$ for CO_2

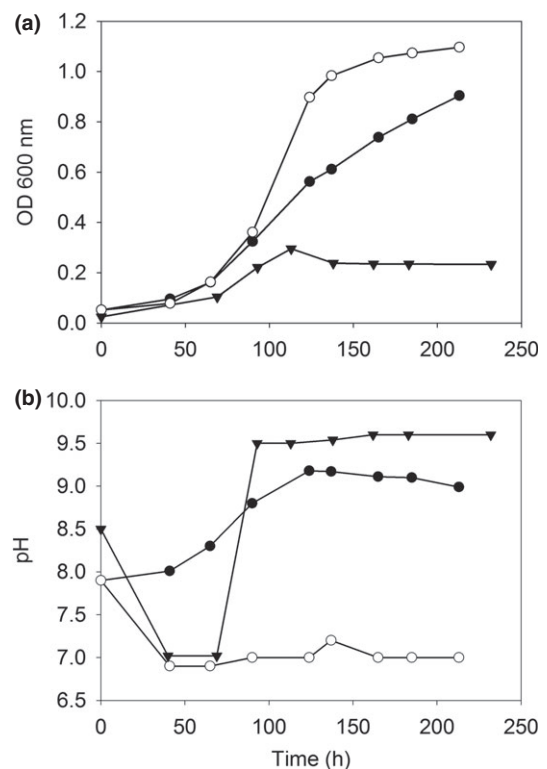


Fig. 1 Growth of *Thalassiosira pseudonana* and pH of culture at 400 ppm CO_2 (closed circles), 20 000 ppm CO_2 (open circles) and then switched from 20 000 to 50 ppm CO_2 after 60 h (triangles). (a) Growth followed using optical density at 600 nm; (b) culture pH.

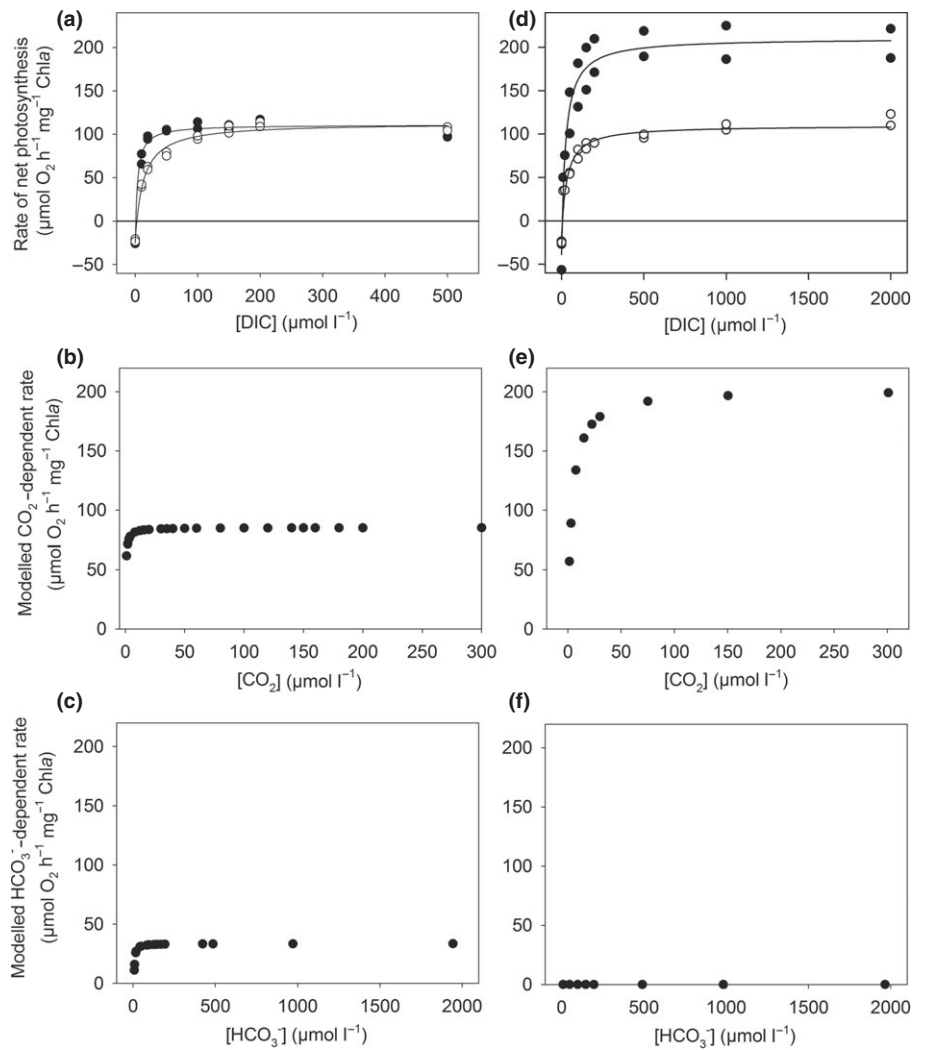


Fig. 2 Rate of net photosynthesis of *Thalassiosira pseudonana* grown at 400 or 20 000 ppm CO₂. (a) Rate measured for 400 ppm cultures at pH 7 (closed circles) or pH 8 (open circles) vs concentration of dissolved inorganic carbon (DIC). (b) Modelled rate for 400 ppm cultures for combined pH values vs concentration of CO₂. (c) Modelled rate for 400 ppm cultures for combined pH values vs concentration of HCO₃⁻. (d) Rate measured for 20 000 ppm cultures at pH 7 (closed circles) or pH 8 (open circles) vs concentration of dissolved inorganic carbon. (e) Modelled rate for 20 000 ppm cultures for combined pH values vs concentration of CO₂. (f) Modelled rate for 20 000 ppm cultures for combined pH values vs concentration of HCO₃⁻. The kinetic parameters for the model are shown in Table 2.

Table 1 Kinetics of photosynthesis by *Thalassiosira pseudonana* grown at different CO₂ concentrations and measured at different pH values and treated with 0.4 mM acetazolamide (AZA)

CO ₂ (ppm)	pH	V _{net} ^{max} (μmol O ₂ h ⁻¹ mg ⁻¹ Chla)	K _{1/2} (μmol DIC l ⁻¹)	CP (μmol DIC l ⁻¹)	Slope (μmol O ₂ h ⁻¹ mg ⁻¹ Chla μmol ⁻¹ DIC l)	R ²
400	7	111 (3)	4.2 (0.9)	0.8 (0.2)	26 (6)	0.99
400	8	113 (3)	15.3 (1.6)	2.4 (0.4)	7 (1)	0.99
20 000	7	205 (17)	58.9 (22.6)	2.9 (5.9)	4 (1)	0.74
20 000	8	95 (8)	46.5 (19)	5.1 (4.7)	2 (1)	0.71
50 (6 h)	7	179 (5)	25.3 (3.6)	3.1 (0.9)	7 (1)	0.95
50 (12 h)	7	156 (3)	13.3 (1.4)	1.8 (0.4)	12 (1)	0.97
400 + AZA	7	163 (7)	23.0 (4.2)	3.2 (0.9)	7 (1)	0.96

Values are means with standard error in parentheses.

DIC, dissolved inorganic carbon; V_{net}^{max}, the maximum rate of net photosynthesis; K_{1/2}, half-saturation concentration for DIC; CP, compensation point.

was nearly identical to that of DIC and the contribution of HCO₃⁻ was zero (Fig. 2e,f; Table 2). The half-saturation concentration for CO₂ was 3.8 μM and the compensation point was again close to 0 for CO₂. In comparison to the cells grown at 400 ppm, cells at 20 000 ppm had a 1.8-fold greater V_{net}^{max}, a

nearly 10-fold higher half-saturation constant for CO₂, and thereby a 5.5-fold lower slope against CO₂ and lacked the ability to use HCO₃⁻. At ambient conditions, presumed to be 16 μM CO₂ and 2000 μM HCO₃⁻, the rate of net photosynthesis was 98% saturated for cells grown at 400 ppm but only 80%

saturated for cells grown at 20 000 ppm (Table 2). For a 10-fold lower CO₂ concentration of 1.6 μM, the rates of net photosynthesis were 81% and 30% saturated for cells grown at 400 and 20 000 ppm, respectively. Ambient concentrations of 2000 μM HCO₃⁻ were saturating for cells grown at 400 ppm, but HCO₃⁻ use was absent in cells grown at 20 000 ppm.

Net photosynthetic rate was also measured at pH 7 for cells switched from 20 000 ppm CO₂ to low CO₂ (50 ppm) for 6 or 12 h (Fig. 3). After 6 or 12 h at a low CO₂ concentration, the slopes were lower than that of cells grown at 400 ppm CO₂ concentration. However at pH 7, $V_{\text{net}}^{\text{max}}$ values were intermediate to those found at 400 and 20 000 ppm, with a tendency to decrease as a function of time (Table 1). The half-saturation constant values also decreased as a function of time (Table 1). However, even after 12 h at 50 ppm, the slope of photosynthesis rate to concentration of DIC was lower than for cells grown for several days at 400 ppm.

Enzyme activities

Enzymes that could be involved in biochemical or biophysical CCMs in *T. pseudonana* were studied. The activity of Rubisco was lower in cells grown at 400 ppm than in those grown at 20 000 ppm (1.59-fold, Student's *t*-test, $P < 0.001$; Fig. 4). The rates of Rubisco activity (as carbon) cannot account for the oxygen-based rates of photosynthesis (6 vs 100 μmol h⁻¹ mg⁻¹ Chl*a* at 400 ppm and 20 vs 205 μmol h⁻¹ mg⁻¹ Chl*a* at 20 000 ppm). The activity of fully CO₂-activated Rubisco was, however, approx. threefold higher than that of nonactivated enzyme, at both 400 (18 μmol h⁻¹ mg⁻¹ Chl*a*) and 20 000 ppm CO₂ (60 μmol h⁻¹ mg⁻¹ Chl*a*), but again this was lower than the oxygen-based rates of photosynthesis even after assuming a photosynthetic quotient of 1.26 (Spilling *et al.*, 2015). However, other mechanisms such as activation by protein–protein interaction with, for instance, CbbX may also be involved (Mueller-Cajar *et al.*, 2011). Surprisingly, activities of the C₄ enzymes PEPC and PPDK were also lower (5.3-fold, Student's *t*-test, $P < 0.001$, and 4.6-fold, Student's *t*-test, $P < 0.001$, for PEPC and PPDK, respectively) in cells from 400 ppm than those from 20 000 ppm CO₂. By contrast, in cells grown at 400 ppm, NAD-ME and CA activities were higher (4.3-fold, Student *t*-test,

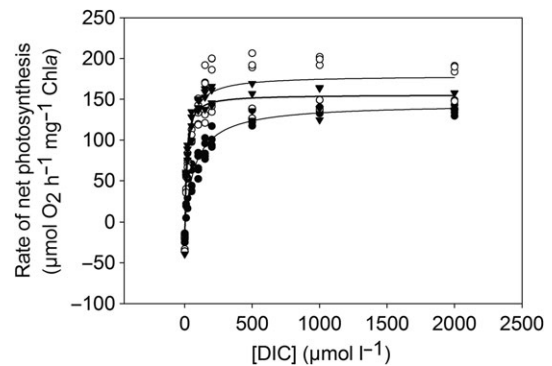


Fig. 3 Rate of net photosynthesis at pH 7 of *Thalassiosira pseudonana* grown at 20 000 ppm CO₂ (closed circles) and then switched to 50 ppm CO₂ for 6 h (open circles) or 12 h (triangles). The experimental data were fitted to a slightly modified Michaelis–Menten equation that took into account the compensation point for dissolved inorganic carbon (DIC); parameters are given in Table 1.

$P < 0.001$, and 3.75-fold, Student *t*-test, $P < 0.001$, respectively) than in cells grown at 20 000 ppm CO₂.

Thalassiosira pseudonana cells acclimated to 20 000 ppm CO₂ were shifted to 50 ppm CO₂ to determine the rate of acclimation and to characterize the CCM under more carbon-limiting conditions. Rubisco and PEPC activities both decreased exponentially with a time constant of 0.086 (0.044) and 0.064 (0.035) h⁻¹, respectively (Fig. 5a). Consequently, the PEPC:Rubisco ratio, which began at *c.* 0.27, also decreased exponentially to reach *c.* 0.07 at 48 h after the switch to low CO₂ (Fig. 5b). Therefore, the PEPC:Rubisco ratios are always much less than 1. At 12 h after the switch to 50 ppm CO₂, the activity of NAD-ME increased 5.6 (1.2)-fold, while that of PPDK decreased 2.4 (0.2)-fold. These data therefore do not support a role for C₄-type photosynthesis in the carbon assimilation of *T. pseudonana*.

By contrast, upon the switch to low CO₂ concentration, CA activity was induced rapidly. The CA activity that was < 300 WAU increased exponentially to reach a modelled value of 4890 (700) WAU (16-fold increase) with a time constant of 0.13 (0.0587) h⁻¹ (Fig. 5c). A ratio between iCA and eCA of *c.* 1 was obtained for cells grown at all three CO₂ concentrations. The effect of inhibiting eCA on the net photosynthetic rate of

Table 2 Modelled kinetics of CO₂-dependent and HCO₃⁻-dependent photosynthesis by *Thalassiosira pseudonana* grown at 400 or 20 000 ppm CO₂

CO ₂ (ppm)	$V_{\text{net}}^{\text{max}}$ (μmol mg ⁻¹ Chl <i>a</i> h ⁻¹)		$K_{1/2}$ (μmol l ⁻¹)		CP (μmol l ⁻¹)		Slope (μmol O ₂ h ⁻¹ mg ⁻¹ Chl <i>a</i> μmol ⁻¹ l)		% of $V_{\text{net}}^{\text{max}}$ under ambient		R^2
	CO ₂	HCO ₃ ⁻	CO ₂	HCO ₃ ⁻	CO ₂	HCO ₃ ⁻	CO ₂	HCO ₃ ⁻	CO ₂	HCO ₃ ⁻	
400	85 (9)	27 (9)	0.4 (0.1)	2.7 (0.4)	0.0 (0.0)	7.5 (0.7)	296 (38)	42 (6)	98	100	0.92
20 000	202 (39)	0.0	3.8 (0.1)	–	0.0 (0.0)	–	53 (9)	–	60	–	0.60

Values are means with standard error of the estimate in parenthesis. Estimated rates as a percentage of $V_{\text{net}}^{\text{max}}$ are calculated for 16 μM CO₂ and 2000 μM HCO₃⁻. The raw data are shown in Fig. 2(a,d) and the outcomes of the models are shown in Fig. 2(b,c,e,f).

–, not applicable as bicarbonate use is absent.

$V_{\text{net}}^{\text{max}}$, the maximum rate of net photosynthesis; $K_{1/2}$, half-saturation concentration for DIC; CP, compensation point.

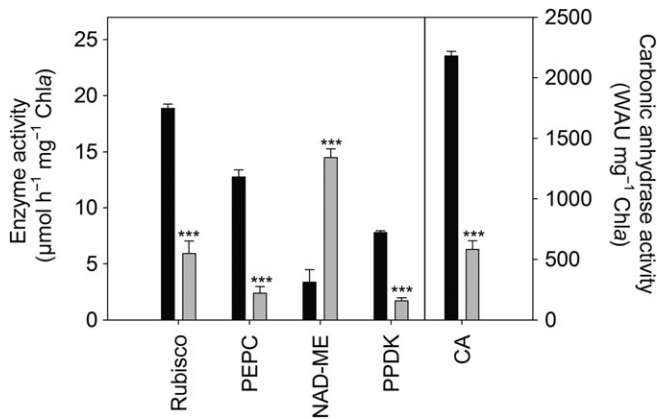


Fig. 4 Activities of partially CO₂-activated Rubisco, C₄ enzymes and carbonic anhydrase (CA) in *Thalassiosira pseudonana* grown at 400 ppm CO₂ (black bars) and 20 000 ppm CO₂ (grey bars). Bars to the left-hand side of the vertical line refer to the left-hand axis, and vice versa. Error bars represent + 1 SD. ***, *P* < 0.001. WAU, Wilbur-Anderson Units; PEPC, phosphoenolpyruvate carboxylase; NAD-ME, NAD-dependent malic enzyme; PPDK, pyruvate phosphate dikinase.

the cells grown at 400 ppm CO₂ was tested at pH 7 using a specific inhibitor of eCA, AZA. The addition of AZA increased the half-saturation constant for DIC fivefold, increased the compensation point *c* threefold and decreased the slope of response to DIC fourfold, but did not affect the maximum rate of net photosynthesis (Fig. 6; Table 1). The kinetic response of cells grown at 400 ppm CO₂ but treated with AZA resembled those grown at 20 000 ppm CO₂ (Table 1), suggesting that eCA has a key role in the carbon uptake properties of *T. pseudonana*.

In order to check if the response to low CO₂ was reversed on return to high CO₂, cultures grown at 20 000 ppm and switched to 50 ppm for 24 h were then switched back to 20 000 ppm for 12 h. While PEPC and PPDK activity increased (by 4.5-fold, Student's *t*-test, *P* < 0.001, and 5.3-fold, Student's *t*-test, *P* < 0.001, respectively), Rubisco activity did not change. NAD-ME and CA activity decreased (1.4-fold, Student's *t*-test, *P* < 0.05, and 3.3-fold, Student's *t*-test, *P* < 0.001, respectively) (Fig. 7).

These results show that the responses of *T. pseudonana* to CO₂ are rapid and reversible. The kinetic properties of carbon uptake are strongly linked to the activity of CA, and the enzyme activity profiles suggest that carbon fixation involves C₃ rather than C₄ metabolism.

Discussion

Biophysical CCM in *T. pseudonana*

Cells grown at 400 ppm CO₂ have a *K*_{1/2} for CO₂ of only 0.4 µM, which is in good agreement with the growth *K*_{1/2} estimated for *T. pseudonana* (Clark & Flynn, 2000) at 273 µM DIC, equivalent to *c*. 1.4 µM CO₂ under their experimental conditions. Both estimates are substantially lower than the *K*_{1/2} for diatom 1D Rubisco (20–60 µM) (Whitney *et al.*, 2011), which clearly indicates that some form of CCM is operating. *T. pseudonana* grown

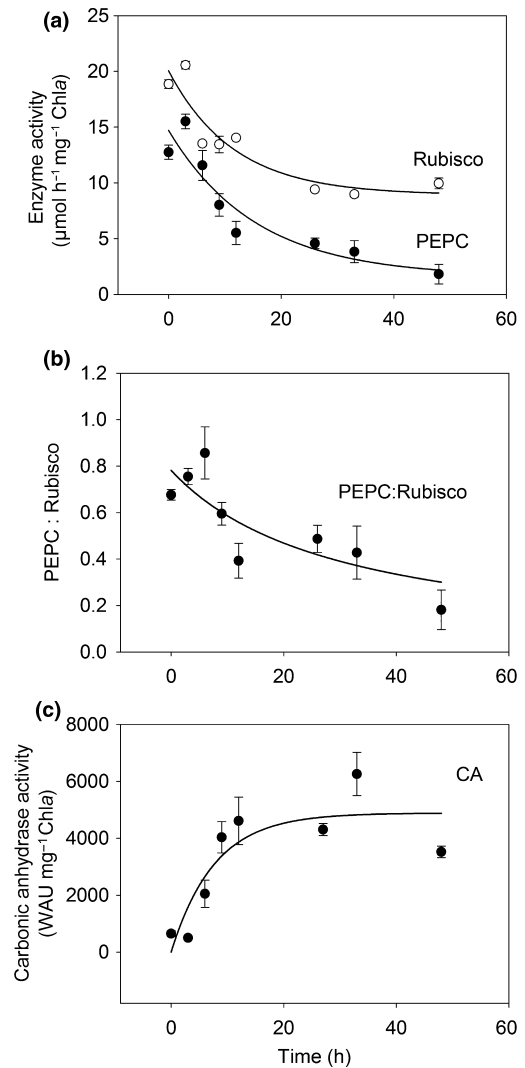


Fig. 5 Time course of enzyme activities after switching cultures from 20 000 ppm to 50 ppm CO₂. (a) Activities of partially CO₂-activated Rubisco (open circles) and PEP carboxylase (PEPC; closed circles). (b) Ratio of PEPC : Rubisco. (c) Activity of carbonic anhydrase (CA). Error bars represent ± 1 SD. WAU, Wilbur-Anderson Units.

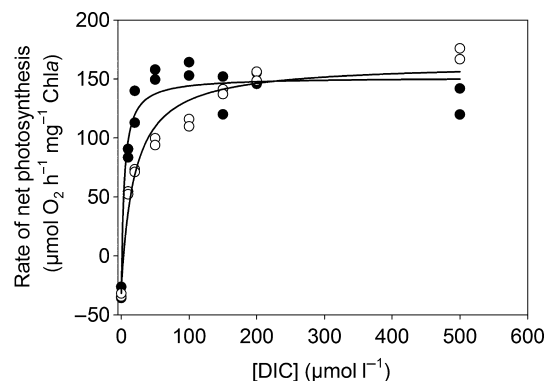


Fig. 6 Effect of acetazolamide (AZA; 0.4 mM) on the kinetics of carbon uptake at pH 7 for *Thalassiosira pseudonana* grown at 400 ppm CO₂. Control (closed circles) and treated cells (open circles) are shown.

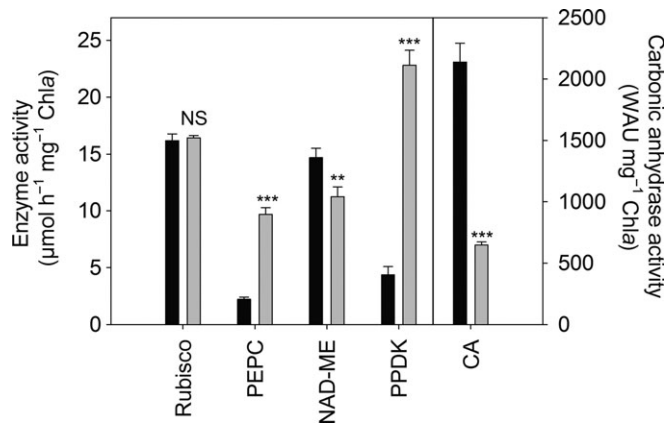


Fig. 7 Activities of partially CO₂-activated Rubisco, C₄ enzymes and carbonic anhydrase (CA) in *Thalassiosira pseudonana* grown at 20 000 ppm CO₂ and switched to 50 ppm CO₂ for 24 h (black bars) and then returned to 20 000 ppm CO₂ for 12 h (grey bars). Bars to the left-hand side of the vertical line refer to the left-hand axis and vice versa. Error bars represent + 1 SD. NS, not significant; **, $P < 0.01$; ***, $P < 0.001$. WAU, Wilbur-Anderson Units; PEPC, phosphoenolpyruvate carboxylase; NAD-ME, NAD-dependent malic enzyme; PPDK, pyruvate phosphate dikinase.

at 400 ppm CO₂ preferentially used CO₂ (70%) rather than HCO₃⁻ (30%) at ambient and saturating conditions despite the *c.* 120-fold higher concentration of HCO₃⁻. This is similar to *P. tricornutum* (Burkhardt *et al.*, 2001) but different from *Thalassiosira weissflogii*, which took up CO₂ and HCO₃⁻ at a similar rate (Burkhardt *et al.*, 2001). For cells grown at 400 ppm, our reported $K_{1/2}$ for DIC at pH 8 and 16°C is very similar to that obtained for the same species at pH 8.2 and 20°C (Nakajima *et al.*, 2013) and to that reported for low CO₂-grown *Chlamydomonas reinhardtii* cells (Sültemeyer *et al.*, 1988).

Cells of *T. pseudonana* grown at a 20 000 ppm CO₂ were only able to use CO₂ and the affinity ($K_{1/2}$) for DIC was over fivefold lower than for cells grown at 400 ppm CO₂, a down-regulation that has been reported in this and other marine diatoms (e.g. Burkhardt *et al.*, 2001; Trimborn *et al.*, 2009; Nakajima *et al.*, 2013) and *C. reinhardtii* (Sültemeyer *et al.*, 1988). In *T. pseudonana*, the $K_{1/2}$ for CO₂ was still *c.* 4 µM and so substantially lower than the $K_{1/2}$ value for Rubisco: some down-regulated form of CCM, therefore, still seems to be operating in *T. pseudonana* grown at 20 000 ppm. External and internal CA activity was also still present in these cells, which might be adequate to promote CO₂ uptake – this is consistent with the finding that some forms of CA are constitutive in this species (Samukawa *et al.*, 2014).

Carbonic anhydrase appears to be crucial in this CCM and that of other marine diatoms (Hopkinson *et al.*, 2011). Our enzymatic activity data showing a rapid fourfold up-regulation at 400 vs 20 000 ppm are similar to previous reports (Hopkinson *et al.*, 2013) and also in agreement with data obtained at the transcriptional level (McGinn & Morel, 2008; Kustka *et al.*, 2014; Samukawa *et al.*, 2014). All these reports indicate an increase in CA under low CO₂. In *T. pseudonana*, CA is present in the periplasmic space, cytosol, mitochondria, periplastidial compartment and stroma (Tanaka *et al.*, 2005; Samukawa *et al.*, 2014).

Using AZA we observed a decreased affinity for DIC, with kinetics very similar to those of cells growing at 20 000 ppm, underlining the important role that extracellular CA plays in this CCM. As far as we are aware, there is no literature for aquatic (or terrestrial) photoautotrophs with C₄ metabolism relying on eCA. On the contrary, work by Reiskind *et al.* (1988) on the CCM in the marine green macroalgae *Udotea flabellum*, which has a C₄ fixation pathway based on PEP carboxykinase, showed that CA was not involved, as the CCM was active in the presence of a CA inhibitor. Furthermore, although the model of Kustka *et al.* (2014) reported up-regulation of a number of carbonic anhydrases, including CA-6 that could be located at the cell surface, in their model (Fig. 6) they located it in the chloroplast endoplasmic reticulum. However, although there is no experimental evidence for eCA being involved in C₄ metabolism, and most interpretations of CA increases are linked to the operation of a biophysical CCM, it is theoretically possible that an eCA could facilitate the rate of inward diffusion of either CO₂ or HCO₃⁻, or both.

As has been found for higher plant Rubisco (Lorimer *et al.*, 1976), we observed an increase of Rubisco activity upon CO₂ activation, presumably linked to a change in Rubisco carbamylation state, which was threefold for cells grown at 400 and 20 000 ppm CO₂. Rubisco activity, whether the enzyme was activated or not, was always higher at high vs low CO₂, on a Chla basis. It is possible that the greater Rubisco activity at high CO₂ increased the capacity to fix CO₂, as there appears to be little excess carboxylation capacity in diatoms (Glover & Morris, 1979).

Evidence for and against C₄ metabolism in *Thalassiosira pseudonana*

Whether or not C₄ photosynthesis is involved in any of the kinetic characteristics that have been observed in *T. pseudonana* has been a matter of debate. Kustka *et al.* (2014) produced a model of C₄ metabolism for *T. pseudonana* in which PEPC, in the chloroplastic endoplasmic reticulum or the periplastidial space, fixes HCO₃⁻ to produce oxaloacetic acid that is transported to the chloroplast where it is decarboxylated by pyruvate carboxylase to produce CO₂ in the vicinity of Rubisco.

In our experiments, the activity of PEPC was lower in cells from low CO₂ (grown at 400 or switched to 50 ppm) than in those from high CO₂ (20 000 ppm): the opposite of what is expected for C₄ metabolism. The ratio of PEPC:Rubisco was also lower at 400 than at 20 000 ppm and decreased with time when cells were switched from 20 000 to 50 ppm. Furthermore, the ratio of PEPC:Rubisco in *T. pseudonana* was always much less than 1, while in aquatic C₄ plants it is between 1.8 and 6.6, and in terrestrial plants it is more than 5 (Zhang *et al.*, 2014). Moreover, the activity of other enzymes required for the operation of the C₄ cycle, such as PPDK, was also lower at low CO₂. Although NAD-ME, one of the three possible decarboxylating C₄ enzymes, had a fourfold higher activity at 400 than at 20 000 ppm, this enzyme also contributes to the overall regulation of malate metabolism in many organisms and thus its

increase in activity is not necessarily associated with C_4 metabolism. Malate is an important substrate for mitochondria, and a significant fraction of glycolytic products enter the Krebs cycle via the combined action of PEPC, malate dehydrogenase, and malic enzyme without any link to C_4 metabolism. Recently it has been shown that NAD-ME is located within the mitochondria in *P. tricornutum* (Xue *et al.*, 2015) and within the cytosol in *T. pseudonana* (Tanaka *et al.*, 2014). This suggests that the CO_2 released from this decarboxylation would not be in the vicinity of Rubisco. Overall, these enzyme activities, and their pattern of change, are inconsistent with the operation of C_4 photosynthesis in this species.

The conclusion that C_4 metabolism is not an important component of the CCM in *T. pseudonana* is in agreement with recent work of Tanaka *et al.* (2014), who observed a greater abundance of PEPC1 and PEPC2 transcripts in high than in low CO_2 . Similarly, the transcripts for other enzymes potentially involved in C_4 photosynthesis, phosphoenolpyruvate carboxykinase (PEPCK), PPK and NAD-ME, were not higher when *T. pseudonana* was grown in low vs high CO_2 , nor were they regulated by the circadian cycle, suggesting that they are not involved in C_4 photosynthesis. The absence of C_4 metabolism was also concluded from the lack of change in PEPC:Rubisco ratio in cells of *T. pseudonana* grown at 50 or 800 ppm (Trimborn *et al.*, 2009). Finally, pulse-chase experiments showed that *T. pseudonana* did not incorporate 4-carbon molecules during photosynthesis and immunoblots showed no difference in PEPC abundance in cells grown at 380 or 100 ppm (Roberts *et al.*, 2007b). By contrast, the addition of 3,3-dichloro-2-(dihydroxyphosphinoylmethyl)-propionate (DCDP), an inhibitor of PEPC, or 3-mercaptopicolinic acid (3-MPA), an inhibitor of PEPCK, reduced photosynthetic activity in *T. pseudonana* (McGinn & Morel, 2008). However, it has subsequently been shown that both inhibitors had no effect on the half-saturation constant, but instead inhibited V_{max} , suggesting that they had a general toxic effect on metabolism rather than a specific effect on the CCM (Tanaka *et al.*, 2005, 2014). The reason for these different conclusions is currently unclear. Kustka *et al.* (2014) reported rapid (within 30 min) but transient (returned close to pretransient levels in 90 min) changes in two forms of PEPC transcripts on transfer from pH 7.61 to 8.48. An alternative explanation to PEPC playing a photosynthetic role is that the response is linked to internal pH homeostasis by the production of carboxylic acids. Haimovich-Dayana *et al.* (2013) concluded that *P. tricornutum* lacked C_4 metabolism and proposed that any C_4 -like metabolism is a futile cycle to dissipate light energy rather than to fix carbon and may also play a role in internal pH homeostasis. Although diatoms such as *T. pseudonana* have biophysical pH regulation mechanisms based on a Na^+ -energized plasmalemma (Taylor *et al.*, 2012), a biochemical pH-stat based on PEPC as part of the glycolytic pathway may also be involved in pH regulation (Sakano, 1998). The steady-state up-regulation of PEPC reported by Kustka *et al.* (2014) of between 1.52- and 1.75-fold is much lower than for the different forms of CA whose protein-level up-regulation is in broad agreement with our changes in activity. Kustka *et al.* (2014) also reported an up-regulation of two forms

of the anion channel Bestrophin (Hartzell *et al.*, 2008) of between 3.31- and 4.24-fold, which could be involved in facilitating diffusion of oxaloacetate into the chloroplast. However, Bestrophin can also act as an HCO_3^- channel (Qu & Hartzell, 2008), which would also be consistent with a biophysically based CCM.

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Author contributions

R.C., S.C.M. and B.G. planned and designed the research. R.C. and L.D. performed the experiments. R.C., S.C.M. and B.G. analysed the data and wrote the manuscript.

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